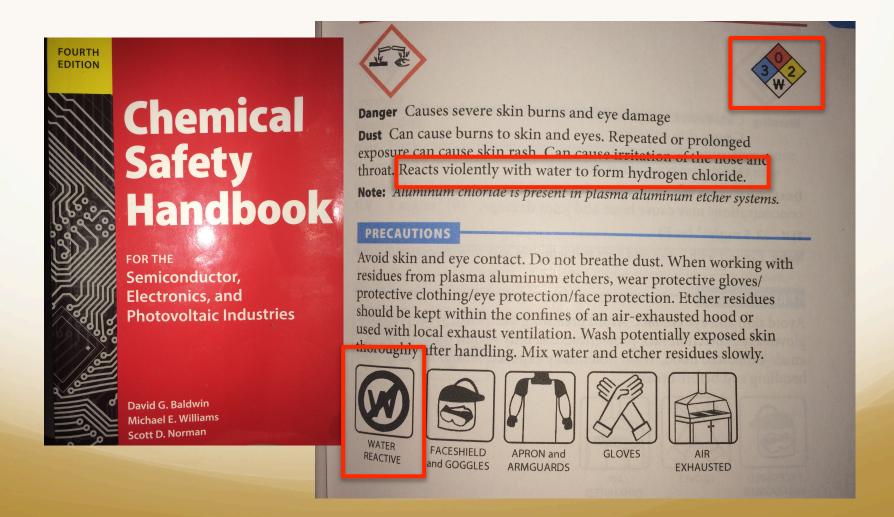
• Incident summary:

- A student received aluminum chloride from Sigma for an experiment.
- The SDS said to store under inert gas and vent periodically.
- The lab didn't have any easy way to do this, but saw that the chemical was water soluble.
- After checking the SDS and calling Sigma to determine if this was a good plan, the student and EHS rep added the chemical to water.
- The result was a reaction with heat and gas production.

Aluminum chloride – a closer look . . .



GHS Label elements, including precautionary statements

Pictogram



Note: No symbol for reactivity. . .

Signal Hazards not otherwise classified (HNOC) or not covered by GHS Reacts violently with water.

Conditions for safe storage, including any incompatibilities.

Keep container tightly closed in a dry and well-ventilated place.

Store under inert gas. Vent periodically. Handle and open container with care. Keep in a dry place.

Storage class (TRGS 510): Non-combustible, acute toxic Cat.3 / toxic hazardous materials or hazardous materials

causing chronic effects

10.	STABIL	_ITY	AND	REACTIVITY	

10.1 Reactivity

No data available

10.2 Chemical stability Stable under recommended storage conditions.

NFPA Rating

Health hazard: 3

Fire Hazard: 0

Reactivity Hazard: 2

Possibility of I Unstable or may react violently if mixed with water.

10.4 Conditions to avoid Avoid moisture.

Reactivity Hazard:

10.5 Incompatible materials

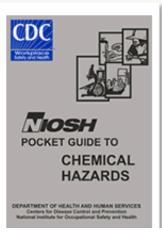
Strong oxidizing agents, Alcohols, Mixtures of nitrobenzene and aluminum chloride are thermally unstable and may lead to explosive decomposition due to a multi-step decomposition reaction occurring above 90 degrees C, which self-accelerates with high exothermicity producing azo- and azoxypolymers.

- What went right?
 - Lab members had attempted to determine if the chemical was water reactive.
 - The procedure was done in a fume hood.
 - The EHS rep reached out to the Coordinator for assistance following the reaction.
- What went wrong?
 - The chemical was water reactive, but the SDS didn't include this in the reactivity section.
 - A more careful reading of the SDS, or a Google search, would have shown that aluminum chloride is water reactive, generating hydrogen chloride.

- Quick guide to SDSs what to look for.
 - Pictograms
 - Other hazards
 - Reactivity
 - NFPA ratings
- Other resources
 - NIOSH guide
 - Chemical compatibility tool Cameo Chemicals

NIOSH Pocket Guide to Chemical Hazards

The NIOSH Pocket Guide to Chemical Hazards is intended as a source of general industrial hygiene information on several hundred chemicals/classes for workers, employers, and occupational health professionals. The NIOSH Pocket Guide does not contain an analysis of all pertinent data, rather it presents key information and data in abbreviated or tabular form for chemicals or substance groupings (e.g. cyanides, fluorides, manganese compounds) that are found in the work environment. The information found in the NIOSH Pocket Guide should help users recognize and control occupational chemical hazards.



Search the NIOSH Pocket (

International Chemical Safety Cards (ICSC)







Enter search terms separated by

The International Chemical Safety Cards (ICSC) summarize essential health and safety information on chemicals for their use at the "shop floor" level by workers and employers in factories, agriculture, construction and other work places.

ICSC summarize health and safety information collected, verified, and peer reviewed by internationally recognized experts, taking into account advice from manufacturers and Poison Control Centres. More about the ICSC.

http://www.cdc.gov/niosh/

http://cameochemicals.noaa.gov/my

Hazard Predictions (for pairs of substances)

ALUMINUM CHLORIDE, ANHYDROUS mixed with WATER

Incompatible

- Reaction products may be corrosive
- · Reaction products may be flammable
- · Reaction liberates gaseous products and may cause pressurization
- · Exothermic reaction at ambient temperatures (releases heat)
- May produce the following gases:
 - Acid Fumes
 - Hydrogen

Documentation

When in doubt, seek help!